

Carbonic Anhydrase-Catalyzed Isotope Equilibrium for Oxygen Isotope Analyses of Aqueous Samples



This patent pending method reduces oxygen isotope equilibrium reaction time from 24h to <1h.

Abstract

Oxygen isotope studies of aqueous samples are typically determined through the isotopic equilibration of CO₂ gas and H₂O. This equilibration process typically takes 24 hours in a closed undisturbed reaction vessel, limiting laboratory throughput.

The technique developed at McMaster utilizes carbonic anhydrase (CA) to rapidly catalyze the oxygen isotope exchange reaction, bringing the reaction time from 24 hours to less than one hour. This technique is compatible with commercially available CO₂-H₂O equilibration devices, and can be easily and immediately adopted by any laboratories that measure the oxygen isotope compositions of aqueous samples.

Applications

Oxygen Isotope analyses in fields such as:

- Environmental sciences: Geology, Hydrology
- Carbonate isotope analyses
- Biology
- Medical

Benefits

- Reduce equilibrium reaction time from 24h to <1h.
- Increase laboratory throughput capability.
- Compatible with existing commercially available equipment.

Tech ID

19-020

Inventors

S. Kim

N. Randazzo

M. Knyf

Patent Status

Patents pending.

Stage of Research

Experimentally verified.
Data is available upon request.

Contact

Alan Sawula
Business Development Officer
905-525-9140 Ext. 28209
sawulad@mcmaster.ca